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Synthesis of the seed germination stimulant 3-methyl-2*H*-furo[2,3-*c*]pyran-2ones utilizing direct and regioselective Ti-crossed aldol addition

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1. Introduction

In 2004, Flematti's group disclosed the unique and remarkable seed germination stimulatory activity of 3-methyl-2H-furo[2,3-*c*]-pyran-2-one **1** in a wide range of plant species.¹ This lead generation finding is considered to be a promising candidate tool for the plant production industry. Since this discovery, three syntheses have appeared: the first synthesis by Flematti's group themselves,² the second by Goddard-Borger's group,³ and the third by Xu's group.⁴ Recently, a patent by one of the author (N.M.) groups was released.⁵

The first method, starting from pyromeconic acid, is short and straightforward but results in low overall yield (ca. <10%). The second method produces a significantly greater total yield to 30%, but requires 9 steps from 1,2-*O*-isopropylidene-D-xylofuranose as the starting compound. The third method is performed in 8 steps from diethyl isopropylidenemalonate in 8% overall yield. Our longstand-





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ABSTRACT

3-Methyl-2*H*-furo[2,3-*c*]pyran-2-ones **1** and **2**, a unique and remarkable seed germination stimulant, and its analogue were synthesized using direct and regioselective Ti-crossed aldol addition between dihydro-2*H*-pyran-3(4*H*)-ones and methyl pyruvate as the key step, followed by furanone formation. © 2008 Elsevier Ltd. All rights reserved.

> ing interest in Ti-Claisen condensations,⁶ relevant Ti-direct aldoltype,⁷ and Mannich-type additions⁸ led us to investigate the synthesis of **1** and its dihydro analogue **2** (Fig. 1). We present herein an efficient short synthesis of **1** and **2** utilizing a direct and regioselective Ti-crossed aldol addition between dihydro-2*H*-pyran-3(4*H*)-ones and methyl pyruvate as the key step, followed by furanone formation.

2. Results and discussion

Scheme 1 outlines the retrosynthetic route of **1** starting from 2furfurylmethanol (**3**). The reported oxidative ring expansion of **3** using *m*CPBA,⁹ followed by treatment with AcCl–Et₃N–cat. DMAP, gave 5,6-dihydro-5-oxo-2*H*-pyran-2-yl acetate (**4**) in 82% yield



Scheme 1. Retrosynthesis of 3-methyl-2H-furo[2,3-c]pyran-2-one 1.

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Scheme 2. Reagents and conditions: (a) *m*CPBA/CH₂Cl₂, 0–5 °C; (b) AcCl–Et₃N–cat. DMAP/CH₂Cl₂, 0–5 °C; (c) ^{*i*}PrOH–cat. SnCl₄/ClCH₂CH₂Cl, 20–25 °C; (d) H₂–cat. Pd–C/ AcOEt, 20–25 °C.



Scheme 3. Regioselective direct Ti-aldol reaction of dihydro-2*H*-pyran-3(4*H*)-ones **5** and **7**.

(Scheme 2).¹⁰ An SnCl₄-catalyzed acetal exchange with ⁱPrOH and subsequent catalytic hydrogenation gave 6-isopropoxy-2*H*-pyran-3(6H)-one (**5**) in 68% yield (2 steps).¹¹

The initial model study of the Ti-crossed direct aldol addition was guided using commercially available dihydro-2*H*-pyran-3(4*H*)-one **6** and methyl pyruvate (Scheme 3). Regioselectivity of the enolate formation of dihydro-2*H*-pyran-3(4*H*)-ones **5** and **6** was speculated to be of primary importance in this synthetic plan. Extensively reported studies on this issue revealed that the desired enolate formation was kinetically as well as thermodynamically predominant.¹² Indeed, Ti-direct aldol addition using TiCl₄-Bu₃N reagent between **6** and methyl pyruvate proceeded smoothly at the C4 position, presumably under kinetic conditions (-55 to -78 °C), to give the aldol adduct **8** as a desired sole regioisomer in 71% yield. A similar result was observed in our previous synthetic study of (*R*)-mintlactone.^{7b} The presence of a TMSCl co-reagent did not affect the present reaction.^{7a}

With this result in hand, the reaction using **5** also successfully afforded the key aldol-adduct **7** in 47% yield. Although the yield was decreased, probably due to the labile acetal moiety in **5**, the regiochemistry was similar to that when **6** was used.

Subsequent furanone transformation was investigated. The direct transformation of **8** using *p*-toluenesulfonic acid (PTS)·H₂O catalyst gave the desired furanone **2** in 48% yield (Scheme 4). The identical reaction using **6**, however, gave disappointing results: only a trace amount of the desired product **1** was detected due to decomposition of the acetal moiety in **7**. To overcome this problem, the stepwise dehydration and lactonization procedures were examined: Successive treatment of **6** with (CF₃CO)₂O–Et₃N–cat. DMAP and DBU gave the conjugated ene-dicarbonyl intermediate **10**, which was treated with a PTS·H₂O catalyst to successfully afford the desired furanone **1** in 45% yield (Scheme 5).

In conclusion, we have achieved a new synthesis of 3-methyl-2*H*-furo[2,3-*c*]pyran-2-ones **1** (7 steps; 12% overall yield) and **2** (2 steps; 34% overall yield) through a direct and regioselective Ti-aldol addition. Further studies of the analogue synthesis and the structure–activity relationship are in progress.



Scheme 4. Direct furanone transformation of 8-2 using PTS-H₂O.



Scheme 5. Reagents and conditions: (a) $(CF_3CO)_2O-Et_3N-cat.$ DMAP/ CH₃CN, 0–5 °C; (b) DBU/CH₃CN, 0–5 °C; (c) PTS·H₂O/ toluene, reflux.

3. Experimental

3.1. General

Melting points were determined on a hot-stage microscope apparatus (Yanagimoto) and were uncorrected. NMR spectra were recorded on a JEOL DELTA 300 spectrometer, operating at 300 MHz for ¹H NMR and 75 MHz for ¹³C NMR. Chemical shifts (δ ppm) in CDCl₃ were reported downfield from TMS (=0) for ¹H NMR. For ¹³C NMR, chemical shifts were reported in the scale relative to CDCl₃ (77.00 ppm) as an internal reference. IR Spectra were recorded on a SHIMADZU FT/IR-8100A spectrophotometer.

3.1.1. 5,6-Dihydro-5-oxo-2H-pyran-2-yl acetate (4)^{8,10}

mCPBA (18.1 g, 105 mmol) was added to a stirred solution of 2-furfurylmethanol (3; 6.87 g, 70.0 mmol) in CH₂Cl₂ (210 mL) at 0-5 °C under an Ar atmosphere, and the mixture was stirred for 6 h at the same temperature. The resultant white precipitates were removed by filtration and the filtrate was concentrated under reduced pressure. The obtained residue was diluted with CH₂Cl₂ (190 mL) and cooled to 0-5 °C. Et₃N (12.0 g, 119 mmol) and DMAP (420 mg, 3.5 mmol) in CH₂Cl₂ (10 mL), and AcCl (9.34 g, 119 mmol) in CH₂Cl₂ (10 mL) were successively added to the stirred solution at 0–5 °C under an Ar atmosphere, and the mixture was stirred for 2 h at the same temperature. The mixture was poured into NaHCO₃ aqueous solution and extracted twice with CH₂Cl₂. The combined organic phase was washed with water, brine, dried (Na₂SO₄), and concentrated. The obtained crude oil was purified by SiO_2 -column chromatography (neutral) (hexane-AcOEt = 4:1) to give the desired product 4 (8.96 g, 82%).

Yellow oil; ¹H NMR (CDCl₃) δ 2.15 (3H, s), 4.22 (1H, d, $J_{gem} = 16.9$ Hz), 4.52 (1H, d, $J_{gem} = 16.9$ Hz), 6.28 (1H, d, J = 10.3 Hz), 6.50 (1H, d, J = 3.9 Hz), 6.94 (1H, dd, J = 3.9, 10.3 Hz); ¹³C NMR (CDCl₃) δ 20.8, 67.3, 86.5, 128.6, 142.2, 169.4, 193.3; IR (neat) 3485, 2988, 1751, 1373, 1221, 1006, 933 cm⁻¹.

3.1.2. Dihydro-6-isopropoxy-2H-pyran-3(4H)-one (5)

SnCl₄ (1.0 M in CH₂Cl₂, 750 μ L, 0.75 mmol) was added to a stirred solution of **4** (2.34 g, 15.0 mmol) and ^{*i*}PrOH (1.80 g, 30.0 mmol) in CH₂Cl₂ (45 mL) at 20–25 °C under an Ar atmosphere, followed by being stirred for 2 h. Satd NaHCO₃ aqueous solution was added to the mixture, which was extracted twice with AcOEt. The combined organic phase was washed with brine, dried (Na₂SO₄), and concentrated. The obtained crude oil was purified by SiO₂-column chromatography (neutral) (hexane–AcOEt = 10:1) to give 6isopropoxy-2*H*-pyran-3(6*H*)-one (1.97 g, 84%).

Yellow oil; ¹H NMR (CDCl₃) δ 1.23 (3H, d, *J* = 6.2 Hz), 1.27 (3H, d, *J* = 6.2 Hz), 4.05 (1H, sept, *J* = 6.2 Hz), 4.09 (1H, d, *J*_{gem} = 16.9 Hz), 4.49 (1H, d, *J*_{gem} = 16.9 Hz), 5.31 (1H, d, *J* = 3.4 Hz), 6.13 (1H, d, *J* = 10.3 Hz) 6.86 (1H, dd, *J* = 3.4, 10.3 Hz); ¹³C NMR (CDCl₃) δ 21.7, 23.2, 66.2, 71.0, 91.3, 127.6, 145.0, 195.0; IR (neat) 2975, 1709, 1383, 1262, 1181, 1159, 1121, 1102, 1080, 1042 cm⁻¹.

4510

10% Pd–C (670 mg, 0.63 mmol) was added to a stirred solution of 6-isopropoxy-2*H*-pyran-3(6*H*)-one (1.97 g, 12.6 mmol) in AcOEt (25 mL), and the mixture was stirred equipped with a H₂ balloon for 1.5 h at 20–25 °C. The mixture was filtered through the Celite using glass filter, and the filtrate was concentrated under reduced pressure. The obtained crude oil was purified with SiO₂-column chromatography (hexane–AcOEt = 10:1) to give the desired product **5** (1.62 g, 81%).

Colorless oil; ¹H NMR (CDCl₃) δ 1.18 (3H, d, J = 6.2 Hz), 1.23 (3H, d, J = 6.2 Hz), 1.96 (1H, dddd, J = 4.1, 6.2, 7.2 Hz, J_{gem} = 14.1 Hz), 2.25 (1H, dddd, J = 4.1, 5.9, 8.3 Hz, J_{gem} = 14.1 Hz), 2.45 (1H, ddd, J = 6.2, 7.2 Hz, J_{gem} = 16.5 Hz), 2.59 (1H, ddd, J = 5.9, 8.3 Hz, J_{gem} = 16.5 Hz), 3.90 (1H, d, J_{gem} = 16.9 Hz), 3.98 (1H, sept, J = 6.2 Hz), 4.20 (1H, d, J_{gem} = 16.9 Hz), 5.10 (1H, t, J = 4.1 Hz); ¹³C NMR (CDCl₃) δ 21.4, 23.3, 28.6, 33.8, 67.2, 68.9, 93.8, 209.1; IR (neat) 2975, 1736, 1381, 1248, 1200, 1171, 1125, 1076, 1046, 1009 cm⁻¹. HRMS (ESI) calcd for C₈H₁₄O₃ (M⁺) 158.0943, found 158.0938.

3.1.3. Methyl 2-(tetrahydro-5-oxo-2*H*-pyran-4-yl)-2-hydroxypropanoate (8)

TiCl₄ (162 μ L, 1.5 mmol) and a solution of Bu₃N (371 mg, 2.0 mmol) in CH₂Cl₂ (0.5 mL) were successively added to a stirred solution of dihydro-2*H*-pyran-3(4*H*)-one (**6**; 100 mg, 1.0 mmol) in CH₂Cl₂ (4.0 mL) at -78 °C under an Ar atmosphere. After 15 min, methyl pyruvate (204 mg, 2.0 mmol) in CH₂Cl₂ (0.5 mL) was added to the mixture at -78 °C, followed by being stirred for 1 h. Water was added to the mixture, which was extracted twice with AcOEt. The combined organic phase was washed with water, brine, dried (Na₂SO₄) and concentrated. The obtained crude oil was purified by SiO₂-column chromatography (neutral) (hexane-AcOEt = 10:1) to give the desired product **8** (144 mg, 71%).

Diastereomixtures; Colorless oil; ¹H NMR (CDCl₃) δ 1.33 (3H × 3/5, s), 1.46 (3H × 2/5, s), 2.12–2.30 (2H, m), 2.83 (1H × 2/5, dd, *J* = 7.6, 10.7 Hz), 3.12 (1H × 3/5, dd, *J* = 8.3, 10.7 Hz), 3.73–3.88 (1H, m), 3.78 (3 H × 2/5, s), 3.79 (3H × 3/5, s), 3.88–4.13 (3H, m); ¹³C NMR (CDCl₃) δ 23.0, 24.9, 47.5, 53.1, 66.6, 73.0, 175.7, 176.0, 207.3; IR (neat) 3489, 2974, 1748, 1452, 1379, 1184, 1122, 1026, 979 cm⁻¹. HRMS (ESI) calcd for C₉H₁₄O₅ (M⁺) 202.0841, found 202.0939.

3.1.4. Methyl 2-(tetrahydro-2-isopropoxy-5-oxo-2*H*-pyran-4-yl)-2-hydroxypropanoate (7)

TiCl₄ (484 μ L, 4.4 mmol) was added to a stirred solution of dihydro-6-isopropoxy-2*H*-pyran-3(4*H*)-one (**5**; 316 mg, 2.0 mmol), Bu₃N (1.11 g, 6.0 mmol), and methyl pyruvate (408 mg, 4.0 mmol) in CH₂Cl₂ (6.0 mL) at -60--55 °C under an Ar atmosphere, and the mixture was stirred for 1.5 h at the same temperature. A similar work-up in the case of **8** gave the desired product **7** (245 mg, 47%).

Diastereomixtures; Yellow oil; ¹H NMR (CDCl₃) δ 1.15–1.27 (6H, m), 1.29 (3H \times 3/10, s), 1.34 (3H \times 2/10, s), 1.42 (3H \times 4/ 10, s), 1.48 (3H \times 1/10, s), 1.92 (1H \times 2/10, dt, J = 6.5 Hz, J_{gem} = 14.1 Hz), 1.99 (1H × 1/10, ddd, J = 6.2, 12.0 Hz, J_{gem} = 14.5 Hz), 2.11 (1 H \times 4/10, ddd, J = 2.4, 7.2 Hz, J_{gem} = 13.1 Hz), 2.19– 2.34 (1H \times 4/10 + 2H \times 3/10, m), 2.38 (1H \times 1/10, dt, J = 6.2 Hz, J_{gem} = 14.5 Hz), 2.52 (1H × 2/10, dt, J = 6.5 Hz, J_{gem} = 14.5 Hz), 2.92 (1H \times 1/10, dd, J = 5.9, 12.4 Hz), 3.09 (1H \times 4/10, dd, J = 7.2, 12.4 Hz), 3.14 (1H \times 2/10, dd, J = 6.2, 13.8 Hz), 3.38 (1H \times 3/10, dd, J = 7.2, 12.0 Hz), 3.762 ($3H \times 3/10$, s), 3.763 ($3H \times 3/10$) 2/10, s), 3.79 ($3H \times 1/10$, s), 3.81 ($3H \times 4/10$, s), 3.83-4.04(2H, m), 4.10–4.27 (1H, m), 5.11 (1H \times 4/10, t, J = 2.4 Hz), 5.17 (1H \times 3/10, t, J = 2.4 Hz), 5.18 (1H \times 1/10, t, J = 6.2 Hz), 5.25 (1H × 2/10, t, J = 6.5 Hz); ¹³C NMR (CDCl₃) δ 21.2, 21.3, 21.5, 23.0, 23.2, 23.3, 23.36, 23.40, 23.6, 23.9, 27.6, 29.9, 30.6, 32.3, 49.9, 50.1, 50.6, 51.2, 52.6, 52.8, 52.9, 66.3, 66.7, 67.4, 67.9, 68.7, 68.9, 69.2, 69.4, 72.2, 72.8, 74.8, 75.9, 93.5, 93.8, 95.1, 95.5, 175.0, 175.1, 176.8, 177.0, 206.8, 209.3, 210.7, 210.8; IR (neat) 2974, 1736, 1458, 1372, 1333, 1250, 1190, 1127, 1044, 980 cm⁻¹. HRMS (ESI) calcd for $C_{12}H_{20}O_6$ (M⁺) 260.1260, found 260.1258.

3.1.5. 3-Methyl-2*H*-furo[2,3-c]pyran-2-one (1)¹

(CF₃CO)₂O (71 µL, 0.51 mmol) was added to a stirred solution of 7 (78 mg, 0.30 mmol), Et₃N (46 mg, 0.45 mmol), and DMAP (1.8 mg, 0.015 mmol) in CH₃CN (1.0 mL) at 0-5 °C under an Ar atmosphere, followed by being stirred for 30 min. Water was added to the mixture, which was extracted twice with AcOEt. The combined organic phase was washed with water, brine, dried (Na₂SO₄), and concentrated to the product, to which DBU (64 mg, 0.42 mmol) in CH₃CN (1.0 mL) was added at 0-5 °C under an Ar atmosphere, followed by being stirred for 1 h. Satd NaHCO₃ agueous solution was added to the mixture, which was extracted twice with AcOEt. The combined organic phase was washed with water, brine, dried (Na₂SO₄), and concentrated. A solution of the obtained crude product and PTSH₂O (68 mg, 0.36 mmol) in toluene (2 mL) was refluxed for 2 h under an Ar atmosphere, followed by being cooled down to room temperature. Satd NaHCO₃ aqueous solution was added to the mixture, which was extracted twice with AcOEt. The combined organic phase was washed with water, brine, and dried (Na₂SO₄), and concentrated. The obtained crude oil was purified with SiO_2 -column chromatography (hexane-AcOEt = 4:1) to give the desired product 1 (20 mg, 45%).

Yellow crystals; mp 118–119 °C; ¹H NMR (CDCl₃) δ 1.94 (*Me*; 3H, s), 6.52 (4-*H*; 1H, d, *J* = 5.5 Hz), 7.32 (5-*H*; 1H, d, *J* = 5.5 Hz), 7.44 (7-*H*; 1H, s); ¹³C NMR (CDCl₃) δ 7.6, 100.2, 103.4, 126.8, 139.7, 142.2, 148.0, 171.2; IR (KBr) 1734 cm⁻¹. These spectroscopic parameters agreed with those reported.

3.1.6. 4,5-Dihydro-3-methyl-2H-furo[2,3-c]pyran-2-one (2)

A solution of **8** (61 mg, 0.3 mmol) and PTS (*p*-toluenesulfonic acid)·H₂O (114 mg, 0.6 mmol) in toluene (9 mL) was refluxed for 1 h, under an Ar atmosphere, followed by being cooled down to room temperature. Satd NaHCO₃ aqueous solution was added to the mixture, which was extracted twice with AcOEt. The combined organic phase was washed with water, brine, and dried (Na₂SO₄), and concentrated. The obtained crude oil was purified with SiO₂-column chromatography (hexane:AcOEt = 5:1) to give the desired product **2** (22 mg, 48%).

Yellow crystals; mp 71–73 °C; ¹H NMR (CDCl₃) δ 1.89 (3H, s), 2.84 (2H, t, *J* = 6.5 Hz), 4.19 (2H, t, *J* = 6.5 Hz), 6.92 (1H, s); ¹³C NMR (CDCl₃) δ 8.3, 23.0, 66.8, 115.9, 131.6, 138.7, 142.4, 170.8; IR (KBr) 3011, 1779, 1294, 1148, 1067, 878 cm⁻¹. HRMS (ESI) calcd for C₈H₈O₃ (M⁺) 152.0473, found 152.0471.

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References and notes

- (a) Flematti, G. R.; Ghisalberti, E. L.; Dixon, K. W.; Trengove, R. D. Science 2004, 305, 977; (b) Flematti, G. R.; Ghisalberti, E. L.; Dixon, K. W.; Trengove, R. D.; Skelton, W. B.; White, A. H. Aust. J. Chem. 2005, 58, 505.
- Flematti, G. R.; Ghisalberti, E. L.; Dixon, K. W.; Trengove, R. D. *Tetrahedron Lett.* 2005, 46, 5719.
- Goddard-Borger, E. D.; Ghisalberti, E. L.; Stick, R. V. Eur. J. Org. Chem. 2007, 3925.
- Sun, K.; Chen, Y.; Wagerle, T.; Linnstaedt, D.; Currie, M.; Chmura, P.; Song, Y.; Xu, M. Tetrahedron Lett. 2008, 49, 2922.

5. Matsuo, N.; Nagasawa, A.; Mae, M. WO 07 102615.

- 6 (a) Tanabe, Y. Bull. Chem. Soc. Jpn. **1988**, 62, 1917; (b) Tanabe, Y.; Hamasaki, R.; Funakoshi, S. Chem. Commun. **2001**, 1674; (c) Tanabe, Y.; Makita, A.; Funakoshi, S.; Hamasaki, R.; Kawakusu, T. Adv. Synth. Catal. **2002**, 344, 507; (d) Misaki, T.; Nagase, R.; Matsumoto, K.; Tanabe, Y. J. Am. Chem. Soc. **2005**, 127, 2854; (e) lida, A.; Nakazawa, S.; Okabayashi, T.; Horii, A.; Misaki, T.; Tanabe, Y. Org. Lett. **2006**, 8, 5215.
- (a) Tanabe, Y.; Matsumoto, N.; Higashi, T.; Misaki, T.; Itoh, T.; Yamamoto, M.; Mitarai, K.; Nishii, Y. *Tetrahedron* **2002**, *58*, 8269; (b) Tanabe, Y.; Mitarai, K.; Higashi, T.; Misaki, T.; Nishii, Y. *Chem. Commun.* **2002**, 2542; (c) Nagase, R.; Matsumoto, N.; Hosomi, K.; Higashi, T.; Funakoshi, S.; Misaki, T.; Tanabe, Y. *Org. Biomol. Chem.* **2007**, *5*, 151.
- 8. Funatomi, T.; Nakazawa, S.; Matsumoto, K.; Nagase, R.; Tanabe, Y. Chem. Commun. 2008, 771.

- 9. Laliberte, R.; Medawar, G.; Lefebvre, Y. J. Med. Chem. 1973, 16, 1084.
- 10. Instead of the reported method [9] using Ac_2O -pyridine, the use of $AcCl-Et_3N$ cat. DMAP increased the yield (60–70% \rightarrow 82%) and the reaction speed (ca. 10 h \rightarrow 2 h).
- 11. (a) Caddick, S.; Khan, S.; Frost, L. M.; Smith, N. J.; Cheung, S.; Pairaudeau, G. Tetrahedron 2000, 56, 8953; (b) Mucha, B.; Hoffmann, H. M. R. Tetrahedron Lett. 1989, 30, 4489. The reason for the use of *i*-PrOH instead of MeOH for the present method is easier handling with regard to stability and higher boiling point.
- (a) Hirsch, J. A.; Wang, X. L. Synth. Commun. **1982**, *12*, 333; (b) Goldsmith, D. J.; Dickinson, C. M.; Lewis, A. J. Heterocycles **1987**, *25*, 291; (c) Cox, P.; Lister, S.; Gallagher, T. J. Chem. Soc., Perkin Trans. 1 **1990**, 3151.